

In NMR spectroscopy, a diffusion coefficient is determined by fitting or calculation for the following decay curve of signal intensity I(G):

$$\frac{I(G)}{I(0)} = \exp\left[-\left(\gamma G\delta\right)^2 D\left(\Delta - \frac{\delta}{3}\right)\right]$$

 γ : Gyromagnetic ratio / G: Field gradient strength / δ : Field gradient pulse width / Δ : Diffusion time / D: Self diffusion coefficient

A maximum value of Δ is limited by the relaxation time of a sample, while δ is usually the order of ms regardless of NMR systems. Consequently, strong field gradients are required in the case of small diffusion coefficients (e.g. for polymers) and in the case of not so high- γ nuclei as ¹H (e.g. ⁷Li).

< Example >

Measurements of a ⁷Li diffusion coefficient in a solid electrolyte of lithium ion batteries.



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