## HBA TOOL : Delta-family Software for Spinning-Sideband Analysis in Solid-state NMR

In solid-state NMR, MAS (magic-angle spinning) spectra for spin-1/2 dilute nuclei, such as <sup>13</sup>C, exhibit resonance lines reflecting isotropic chemical shifts. On the other hand, when the chemical-shift anisotropy (CSA) is larger than the applied sample-spinning frequency, a series of additional lines, called spinning sidebands (SSBs), may appear. SSBs render the MAS NMR spectra complicated and are unnecessary to obtain isotropic chemical shifts.

By analyzing SSB intensities, however, we can determine the principal values of CSA tensors, which may provide critical information on microscopic structures and dynamics of materials.

"HBA Tool"<sup>†</sup> is an application program which works seamlessly with JEOL Delta software and analyzes SSBs based on Herzfeld-Berger method (*J. Chem. Phys.*, **73**, 6021 (1980).). From the MAS NMR spectra presented on Delta, it is easy and straightforward to obtain CSA information by using "HBA Tool".

Acquisition User	1. When HBA Tool is installed, it is registered in the main window of Delta. At first, click it as shown in the left Figure.
HBA Tool Extract Select File  Dippm Dipm Spinning Speed TpHr	2. Input window of HBA Tool starts. By clicking "Select File" button, the mouse cursor becomes a fingering pointer.
Speed Tolerance     15[Hz]       Intensity     New Window       Simulate SSB     Simulate Powder	Child Date Hamaconic Content 11 al Line 2   File Optimer View Project Blicking Equation Table Analysis Image: Content 2   Image: Content 2 <td< td=""></td<>
3. With the fingering pointer, touch the MAS NMR spectrum peak-picked on Delta. All information of SSB positions and intensities are loaded on Delta.	20 20 42 50 42
<sup>†</sup> Basic parts of HBA Tool were programmed by Dr. Klaus Eichel and his co-worker: HBA 1.5, K. Eichel and R. E. Wasylishen, Dalhousie Univ., Tuebingen, 2006; http://anorganik.uni-tuebingen.de/klaus/soft/ index.php?p=hba/hba	

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## JEOL Application Note

Isotropic Shift	Extrac		4. I (the speciate	nput the isotropic chemical shift e position of the center peak) to ecify a set of the SSB family of erest. Click "Extract" button to
Spinning Speed	[4[kHz]	]	sta	rt SSB analysis.
Speed Tolevance	[15[Hz]	]		
Intensity	New V	Window		
Simulate SS	8 Sin	mulate Powder		
Hersteld-Bereer Analysis EIDIX le Contours Fi Hele Linna III III III IIII IIII IIII IIII III			-0× 2-	5. A window illustrating contour lines appears, informing the end of the analysis.
1.0 0.5 0.0 0.5 1.0 0 0 5 Gooddey				The horizontal and vertical axes of the graph indicate fitting parameters, $\mu$ and $\rho$ , which relate with the principal values of the CSA tensor, $\delta_{11}$ , $\delta_{22}$ , and $\delta_{33}$ : $\mu = (\delta_{11} - \delta_{33})/\nu_r$ , $\rho = (2\delta_{22} - \delta_{11} - \delta_{33})/(\delta_{11} - \delta_{33})$ . The individual contour lines express possible values of $\mu$ and $\rho$ . Their cross point manifests the unique values of $\mu$ and $\rho$ , consistent with all the SSBs. The cross point depicted as a white filled circle indicates the final
				values of $\mu$ and $\rho$ .

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## An example: CSA provides structural information of material.

Figure shows <sup>13</sup>C CPMAS spectrum for calcium formate Ca(HCOO)<sub>2</sub>. Its unit cell involves two crystallographically inequivalent formate ions HCOO<sup>-</sup>; the environments surrounding formate ions are largely different with each other.

However, their <sup>13</sup>C isotropic chemical shifts differ only by 0.7 ppm, not sufficiently reflecting the difference of their environments.



A lot of SSBs exhibit themselves in the CPMAS spectrum observed under slow spinning condition. HBA Tool analyzes the SSB intensities, leading to the principal values of CSA tensors as listed in the above Table; if represented as powder patterns, their difference becomes more evident as shown in the above Figure.

In this way, CSA determined using HBA Tool may yield detailed information on the microscopic structure of materials, whereas isotropic chemical shifts only suggest ambiguous information.

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