## **Application of MICCS-NMR #1**

## Reaction pathway analysis of radical addition

## Reaction:

This note treats triethylborane (Et3B)-mediated radical addition to oxime ether, where a borane complex is proposed as a key intermediate. However, its isolation is very difficult using conventional methods.

Sample is by courtesy of Prof. T. Naito, Prof. O. Miyata, and Dr. M. Ueda of Kobe Pharmaceutical University.

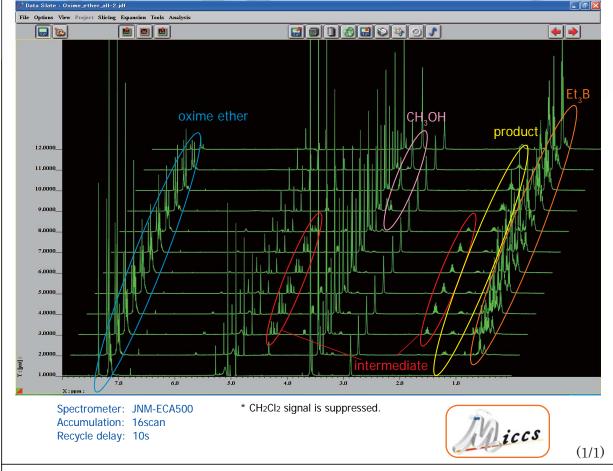
## **Protocol of NMR measurements:**

NMR spectra are observed throughout the reaction, which is initiated by cotroling the flow rate of the individual reactants. When the concentration ratio of oxime ether and Et<sub>3</sub>B becomes 1 to 1, CH<sub>3</sub>OH is added. Thereby, spectra yield signals of reactants, intermediate, and product, sequentially.

1M oxime ether	1M Et <sub>3</sub> B	1M CH <sub>3</sub> OH
10.0	0.5	0.0
9.0	1.0	0.0
8.0	2.0	0.0
7.0	3.0	0.0
6.0	4.0	0.0
5.0	5.0	0.0
5.0	5.0	0.5
4.5	4.5	1.0
4.0	4.0	2.0
3.5	3.5	3.5
2.0	4.0	2.0
1.0	5.0	5.0

\* Concentration of CH<sub>2</sub>Cl<sub>2</sub> solution.

unit : µL/min



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